

# Extraction of Catechins from *Areca catechu* L. Peel with different Solvent Type for Feed Additive of Broiler

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**Abstract**—Catechins are secondary metabolites compound of flavonoids group that is naturally produced by plants and have health benefits as cholesterol-lowering, antioxidant, antimicrobial in mice. Rapid growth in broilers is often followed by high-fat growth as well, thus causing a high cholesterol content in the broiler's meat especially in thigh meat and wings. Feeding extract catechin in broiler as feed additive will reduce cholesterol or fat in broiler's meat. The catechins could be extracted by maceration method, so far there is no information about the type of solvent and extraction method for extraction of catechin from *Areca catechu* peel. The aims of this study to obtain the best combination of solvent type and maceration duration in extracting of catechins from betel nut peel. *Areca catechu* peel used in this study is a green-colored fruit peel, which was harvested from betel nut seeds in Batu Balang village, Lima Puluh Kota District, West Sumatra Province, Indonesia. This study was performed by using Factorial Experimental in Completely Randomized Design with two factors. The first factors was solvent types (water with an initial temperature at 80°C, acetone distillate, and ethyl acetate distillate) and the second factors was maceration duration (6 hours, 12 hours, and 18 hours), each combination treatment was replicated 3 times. The variables observed were percentage of water content, percentage of yield, and percentage of total catechins. The results showed there was a significant interaction ( $P < 0.05$ ) between type of solvent and maceration duration on yield percentage and total catechins percentage, while there was no interaction between type of solvent and maceration duration ( $P > 0.05$ ) on percentage of water content. Both type of solvent and maceration duration were significantly ( $P < 0.05$ ) affected yield percentage and total catechins percentage, while percentage of water content did not affect. It concluded the combination of distillate acetone solvent and maceration duration for 6 hours was the best combination to obtain catechin extract from *Areca catechu* peel. In this condition the percentage of water content was 10.53%, yield percentage was 7.13%, and percentage of catechin extract was 25.53%.

**Keywords**—*Areca catechu* L. peel, catechin extract, feed additive, maceration duration, solvent type.

## I. INTRODUCTION

*Areca catechu* L. is classified as a monocot plant of the Palmae family which is spread almost in all regions in Indonesia. According to the Central Statistics Agency (2017), Areca nut fruit production in Indonesia reached 314.51 tons with an area of 143,927 hectares. Areca nut seed is one of the potential non-oil export commodities in the international market. In 2016 the export volume of betel nuts reached 219,127 tons with a value of US \$ 277.78 million or around Rp. 3.9 billion (Yudha, 2017).

Areca nut fruits are exported in the form of seeds, so that after harvesting the Areca nut seeds will leave the fruit

peel which is a waste and has not been widely used. Areca nut peel waste reaches 76% of the weight of fresh Areca nuts fruits (Mahata *et al.*, 2018). Areca nut peel contains 65.41% water, 34.59% dry matter, 2.22% protein, 0.15% fat, 47.02-53.96% crude fiber, 0.28% Ca, 0.36% P, and energy metabolism 1116 kcal / kg and analysis of betel nut peel with van Soest method showed betel nut peel contains 59.07% Neutral Detergen Fiber (NDF), 44.74% Acid Detergen Fiber (ADF), 27.44% cellulose, 14.32% hemicellulose, and 17, 30% lignin (Mahata *et al.*, 2018). Furthermore, Mahata *et al.* (2018) explained the peel of betel nuts contained phytochemical compounds such as

catechins of 1.466%, total polyphenols 1.693%, total alkaloids 1.382%, and contained fatty acids consisting of 1.83% myristate, palmitate 16.386%, stearate 2.751%, oleic 34.130%, linoleic 2.918%, and linolenic 0.171%.

Catechins was a popular phytochemical compound in Areca catechu plant. Catechins is a metabolic secondary from tannin which produce by plant naturally (Gruenwald *et al.*, 2000). Previous research showed catechins are beneficial for health, as cholesterol-lowering, antioxidants, antimicrobials. Some researcher reported that there was a decrease in the activity of pancreatic cholesterol esterase (pCEase) in vitro in mice given areca seed extract supplements with a reduction in plasma cholesterol by 25%, and did not change triglyceride concentrations (Park *et al.*, 2002). Furthermore, Ikeda (2008) reported that giving catechins from green tea was able to lower total cholesterol levels in rat blood plasma because catechins were able to effectively inhibit the absorption of cholesterol in the intestine. Yunarto *et al.* (2015) also reported, ethyl acetate fraction of gambier leaf extract containing catechins could inhibit the action of HMG-CoA reductase in mevalonate synthesis of HMG-CoA in cells, to reduce total cholesterol, triglycerides, LDL and not increase HDL in rat blood plasma. Raederstorff *et al.* (2003) also reported that the administration of epigallocatechin-gallate from green tea to a dose of 0.7 g/day/kg body weight (BW) was able to reduce LDL and blood plasma of mice. Mechanism of LDL reduction by catechins is reduced by the production of apolipoprotein B which is the main constituent component of LDL and as a radar for LDL receptors in cells (Babu and Liu, 2008).

Broiler is chicken that has high productivity, as meat producer. Broilers have relatively short growth and meat production compared to other livestock. Broiler chicken meat is one of the sources of animal protein that is widely consumed by the community because it is relatively inexpensive, easy to obtain, and has soft fibrous meat so that it is favored by all ages. Muliani (2015) stated the rapid growth of broilers is often followed by high-fat growth as well, thus causing a high cholesterol content in the broiler's body especially in thigh meat and wings. Along with public awareness of health, cholesterol content becomes the public's consideration in consuming broiler meat. The alternative offered is the use of catechins from betel nut peel as additive feeds on broiler rations to lowering cholesterol and fat in broiler meat.

The use of catechins in Areca catechu peel as a feed additive for broiler rations is limited because of high crude fiber content and it still bound in plant cells, so that it is necessary to process by extracting of catechin from Areca

catechu peel. Extraction is the process of separating the material from the mixture using an appropriate solvent. The extraction method is divided into 2 ways, by cold process named maceration method, and by heat process named soxhlet extraction. Maceration is the simplest method of extraction by immersing the sample powder in a suitable solvent at room temperature. Soxhlet extraction is a method of extraction carried out in a device called soxhlet with a polar solvent based on its boiling point. Catechin extraction can be done by the maceration method. The choice of the maceration method in this study because the catechin compound was susceptible to heat, so it was not good to use the soxhlet method (Damanik *et al.*, 2014). This is supported by Cheong *et al.* (2005) that the concentration of catechin compounds decreased in the Soxhlet method compared with the maceration method.

The choice of solvent type is very important when the extraction process. The solvent in maceration extraction process must be in accordance with the material to be extracted, and the solvent must be able to separate quickly after shaking or maceration extraction (Mamonto *et al.*, 2014). Catechins are not soluble in cold water but dissolve in hot water, in alcohol, ethyl acetate, and almost insoluble in chloroform, benzene, and ether (Hidayatullah, 2008). The most efficient solvent for extraction of catechins from green tea is in hot water with temperature at 80°C with maceration duration at 60 minutes and obtained catechin at range 64-97% (Uzunalic *et al.*, 2006). According to Damanik *et al.* (2014) stated the best solvent for extracting catechins from gambier is ethyl acetate concentration 95% with temperature at 60°C, and maceration time at 6 hours with catechin extracted is 87.14%. According to Satriadi (2011) tannin extract on Areca catechu seeds from Pelaihari area in South Kalimantan can be obtained from both water solvent at temperature 80°C with 12 hour maceration duration and acetone solvent with maceration duration for 12 hour, and tannin extracted obtained from both of solvent was 17.97%, and 19.04% respectively.

This study aims was to obtain the best treatment combination between the type of solvent and the best maceration time, to get the optimal catechin extract from the peel of betel nuts as a feed additive broiler ration for lowering broiler blood serum cholesterol.

## II. MATERIAL AND METHOD

### 2.1. Materials Research

The materials used in this study are green betel nut peel collected from Nagari Batu Balang, Lima Pulu Kota District, West Sumatra Province, Indonesia. In addition, aquades, water solvents (with an initial temperature at

80°C), acetone distillate, and ethyl acetate distillate. To measure the catechin content of areca nut peel extract, ethyl acetate pro analysis (pa) 99,5% and catechin standard were used.

The tools used in this research to produce Areca catechu peel flour are oven, grinding machine (ADR MPJ 200), 10 mesh sieve (ABM brand). Areca catechu peel extraction tools used were Erlenmeyer 250 ml volume tubes, aluminum foil, rotary evaporator brand (RV 10 digital V), stirring rods, analytical scales, Whatman 42 filter paper diameter 125 mm, thermometer, incubator shaker (New Brunswick brand). Catechin levels were measured with a UV spectrophotometer, with quartz cuvette equipment, ultrasonic bath, analytical balance, blender, excavator, 50 mm measuring flask, watch glass, petri dish, oven, ordinary filter funnel, 2 ml pipette, Erlenmeyer with a 100 mm sharpener, and qualitative filter paper No.42.

## 2.2. Research Implementation

This study was performed by using Factorial Experimental in Completely Randomized Design with two factors. The first factors was solvent types (water with an initial temperature at 80°C, acetone distillate, and ethyl acetate distillate) and the second factors was maceration duration (6 hours, 12 hours, and 18 hours), each combination treatment was replicated three (3) times.

### 2.2.1. Sample Preparation

Areca catechu peel was chopped and dried using an oven at 60°C until the water content reaches 14%. Then the dried Areca catechu peel is milled by using grinding machine, then sieved with 10 mesh size sieve. Areca catechu peel flour which has been obtained then used for the extraction of catechins by the maceration method with different solvents.

### 2.2.2. Catechin extraction

Catechin was extracted from Areca catechu peel flour by maceration method using different solvent (water with an initial temperature of 80°C, acetone distillate, ethyl acetate distillate) by Uzunalic *et al.* (2006) method. The maceration extraction of Areca catechu peel flour is carried out by weighing 20g of Areca catechu peel flour, and placed into a 250 ml Erlenmeyer, then add with 180 ml of different solvent (water solvent with an initial temperature of 80°C, acetone, and ethyl acetate) in different Erlenmeyer for each solvent. Furthermore, Erlenmeyer was wrapped with aluminum foil to avoid direct sunlight (avoiding light-catalyzed reactions) and they were shaken at incubator shaker for 10 minutes at 60°C, and stirred for all parts of Areca catechu peel flour particles were evenly mixed with solvent, so that extraction can

be carried out perfectly. After 10 minutes, the Erlenmeyer was removed from the shaker incubator and left at room temperature according to different maceration duration treatment (6, 12, and 18 hours), then filtered by using Whatman paper filter number 42. The filtrate obtained was collected in a bottle and wrapped with aluminum foil, then put into a rotary evaporator to vaporize its solvent, so a dry extract of catechins is obtained from each solvent.

## 2.3. Observed variables

### 2.3.1. Percentage of Water Content

The measurement of water content was obtained based on the AOAC method (Association of Official Analytical Chemists, 1990).

### 2.3.2. Percentage of Yield

Extract yield was calculated by ratio of extracted weight produced with the weight of the extracted Areca catechu peel, as below:

$$\% \text{ Yield} = \frac{\text{Extract mass (g)}}{\text{Mass Simplicia (g)}} \times 100\%$$

### 2.3.3. Percentage of Total Catechin

The analysis of the content of the catechin flour of Areca catechu peel is done by SNI method (Indonesian National Standard, 2000).

## III. RESULT AND DISCUSSION

### 3.1. Effect of Solvent Type and Maceration Time on Percentage of Moisture Content of Crude Catechin Extract from *Areca catechu* L.

The average percentage of the water content of crude catechin extract of *Areca catechu* L. peel in each treatment can be seen in Table 1.

Table 1. Mean percentage (%) of the water content of crude catechin extract of *Areca catechu* L. peel based on the solvent type and maceration time.

| Type of Solvent                                | Maceration Time |               |               | Average |
|--|-----------------|---------------|---------------|---------|
|  | M1 (6 hours)    | M2 (12 hours) | M3 (18 hours) |         |
| P1 (water with an initial temperature of 80°C) | 10,95           | 10,91         | 10,85         | 10,90   |
| P2 (acetone distillate)                        | 10,53           | 10,47         | 10,96         | 10,65   |
| P3 (ethyl acetate distillate)                  | 9,93            | 10,17         | 10,93         | 10,35   |
| Average  | 10,47           | 10,52         | 10,91         |         |

Note: ns: not significantly different ( $P > 0.05$ )

SE: Standard error (0.019)

Based on the results of an analysis of variance showed that there was no interaction ( $P > 0.05$ ) between the type of solvent (factor A) and maceration time (factor B) to the percentage of water content from crude catechin extract of *Areca catechu* L. Table 1 shows the percentage of the water content of crude catechin extract of *Areca catechu* L. peel is 9.93 to 10.96%.

Extract water content is the weight of the extract after drying at a temperature of 105 ° C for 30 minutes or after obtaining a constant weight and expressed in percent (Ratnani *et al.*, 2015). Determination of water content aims to provide a minimum limit or the range of the amount of water content in areca nut peel extract. There is no interaction between the type of solvent and the maceration time allegedly because there is no limitation of the time of evaporation of the solvent so that the solvent used at the time of extraction evaporates completely and produces a thick extract with almost the same water content. Prasetyo and Inorlah (2013) reported the process of drying or evaporation of solvents in an extract that is less than optimal can affect the high water content of the extract. This result is different from Damanik (2014) who reported that the highest water content of gambier leaf extract was found in water solvent which was 10.225% and the lowest water content was found in ethyl acetate 95% solvent which was equal to 0.225% with limitation of solvent evaporation time for 60 minutes. The high water content in water is due to the boiling point of the water solvent being higher than ethyl acetate. According to Susanti (2012) states that a solvent must have a boiling point low enough so that the solvent is easily evaporated without using high temperatures in the purification process, ethyl acetate is a type of solvent that has a relatively low boiling point of 77°C so that it is easily evaporated.

DepKes (2008) states that the water content standard of plant extracts in general is <10%, and the water content of catechin extracts from gambir is <14%. Zulharmita *et al.*, (2012) stated the water content of traditional medicine preparations and extracts should not be more than 10%. The water content of EKKBP obtained in this study has met the specified extract water content standard. EKKBP water content in this study ranged from 9.93 to 10.96%, higher than the results of research Ozdemir *et al.* (2018) the water content of catechin extracts in black tea ranged from 4.78 to 5.45%. Yunarto *et al.* (2015) reported the moisture content of the ethyl acetate fraction of gambir leaf extract was 2.23%. However, lower than the results of research Ratnani *et al.* (2015) states that the water content of the bitter leaf extract is 13% exceeds the allowed standard (<10%), causing a high rate of microbial

contamination of  $3.1 \times 10^7$  CFU / g with a standard that is no more than 104 CFU / g. According to Yunarto *et al.* (2017) stated that good storage of catechin extracts <10% because gambier plant catechin compounds are hygroscopic (can draw air water) causing microbial activity is unstable and easily oxidized. The same thing was also reported by Utami (2017) that moisture content exceeding 10% can cause the extract to be easily overgrown with microbes which will reduce the stability of the extract.

### 3.2. Effect of Solvent Type and Maceration Time on Percentage of Total Yield of Crude Catechin Extract from *Areca catechu* L.

The average percentage of total yield of crude catechin extract of *Areca catechu* L. peel in each treatment can be seen in Table 2.

Table 2. Average percentage (%) yield of crude catechin extract of *Areca catechu* L. peel with different types of solvents and maceration times.

| Type of Solvent                                | Maceration Time   |                   |                   | Average           |
|--|-------------------|-------------------|-------------------|-------------------|
|  | M1 (6 hours)      | M2 (12 hours)     | M3 (18 hours)     |                   |
| P1 (water with an initial temperature of 80°C) | 8,20 <sup>a</sup> | 8,14 <sup>a</sup> | 7,94 <sup>b</sup> | 8,09 <sup>a</sup> |
| P2 (acetone distillate)                        | 7,13 <sup>c</sup> | 6,91 <sup>d</sup> | 6,78 <sup>d</sup> | 6,94 <sup>b</sup> |
| P3 (ethyl acetate distillate)                  | 1,44 <sup>e</sup> | 1,38 <sup>e</sup> | 1,37 <sup>e</sup> | 1,40 <sup>c</sup> |
| Average  | 5,59 <sup>a</sup> | 5,48 <sup>b</sup> | 5,36 <sup>c</sup> |                   |

Note: Superscripts with different letters show significantly different effect ( $P < 0.05$ )

Based on the results of the analysis of variance showed that there was an interaction that significantly different effect ( $P < 0.05$ ) between factor A (a type of solvent) and factor B (maceration time) to the percentage of the total yield of crude catechin extract of *Areca catechu* L. peel.

DMRT test results showed that the treatment of P1M1 (water solvent with an initial temperature of 80°C and maceration time 6 hours), and P1M2 (water solvent with an initial temperature of 80°C and maceration time 12 hours) gave a significant effect on the treatment other.

The yield of crude catechin extract of *Areca catechu* L. peel obtained in this study ranged from 1.37% to 8.20%. The highest total yield of crude catechin extract of *Areca catechu* L. peel was obtained from the treatment of water with an initial temperature of 80 ° C with maceration



periods of 6 and 12 hours with values of 8.20% and 8.14%, respectively. The total yield of crude catechin extract of *Areca catechu* L. peel with the highest acetone solvent was obtained at 6 hours maceration time which was 7.13%. While the lowest yield was found in the treatment of ethyl acetate solvents with different maceration times (6, 12, and 18 hours), namely: 1.44%, 1.38%, and 1.37%.

The high yield of EKKBP in the treatment of P1M1, and P1M2 because catechin compounds found in the skin of betel nuts are polar and will dissolve with polar solvents. According to Sayuti (2017), the suitability of the nature of the solvent and the dissolved substance will affect the percentage of yield. Catechin compounds are polyphenol compounds (Yeni *et al.*, 2017), polyphenol compounds are polar (Evans, 2000). According to Tiwari *et al.* (2011), water is a universal polar solvent, and acetone is also classified as a polar solvent, so the yield of EKKBP becomes high in the combination of treatments P1M1, P1M2, followed P1M3, and P2M1. Ethyl acetate is a semi-polar solvent, and is not suitable for dissolving polar catechin compounds, so the total yield of EKKBP is low in the treatment of P3M1, P3M2, and P3M3. Firdiyani *et al.* (2015) stated that ethyl acetate is a solvent with semipolar characteristics.

Types of solvents different in this study showed significant differences in the yield content of EKKBP. The yield content obtained from the water solvent with an initial temperature of 80°C (8.09%) higher than acetone (6.94%) and ethyl acetate (1.40%), and the yield of acetone is higher than ethyl acetate. The high yield of water solvents is due to its nature as a universal solvent and the level of polarity is higher than that of acetone so that more polar substances or compounds from areca nut peels can dissolve, and accumulate into more yields than acetone solvents and ethyl acetate. According to Burdick & Jackson (2012) air is a polar solvent with a polarity index of 10.2, acetone is a polar solvent with a polarity index of 5.1, while ethyl acetate is a semi-polar solvent with a polarity index value of 4.4. Ethyl acetate solvents are classified as semi-polar solvents so they are less able to dissolve polar substances or compounds.

In this study, the longer maceration time (6, 12, and 18 hours) the reduced yield of EKKBP obtained for all types of solvents. The highest yield was obtained at 6 hours maceration (5.59%), then the extension of maceration time was 12 hours, the yield was reduced at 5.48%, and at 18 hours the yield was 5.36%. The low yield due to prolongation of maceration is caused by the process of withdrawal and the amount of compounds from betel nut peel that can dissolve in each solvent has reached a maximum so that the extension time is not much more

compounds left in the peel of betel nut. The results of this study differ from those reported by Kamaluddin (2014), that the longer the extraction time will increase the extract yield. According to Kristian *et al.* (2016) the longer the extraction time, the chance for the material to contact the solvent will be greater, so that the total yield obtained will be high up to the saturation point of the solution, but the number of certain compounds will decrease after reaching the optimal time.

### 3.3. Effect of Solvent Type and Maceration Time on the Percentage of Total Catechins from Betel Nut Peel

The average percentage of total catechins from the peel of betel nut in each treatment can be seen in Table 3.

Table 3. Average percentage (%) of catechins from the extraction of betel nut (*Areca catechu* L.) based on the type of solvent and maceration time

| Type of Solvent                                | Maceration Time    |                    |                    | Average            |
|--|--------------------|--------------------|--------------------|--------------------|
|  | M1 (6 hours)       | M2 (12 hours)      | M3 (18 hours)      |                    |
| P1 (water with an initial temperature of 80°C) | 5,60 <sup>c</sup>  | 4,65 <sup>c</sup>  | 4,46 <sup>c</sup>  | 4,90 <sup>b</sup>  |
| P2 (acetone distillate)                        | 25,53 <sup>a</sup> | 16,89 <sup>b</sup> | 17,17 <sup>b</sup> | 19,87 <sup>a</sup> |
| P3 (ethyl acetate distillate)                  | 23,68 <sup>a</sup> | 17,68 <sup>b</sup> | 17,90 <sup>b</sup> | 19,75 <sup>a</sup> |
| Average  | 18,27 <sup>a</sup> | 13,07 <sup>b</sup> | 13,18 <sup>b</sup> |                    |

Note: Superscripts with different letters show significantly different effect ( $P < 0.05$ )

Based on the results of an analysis of variance showed that there was an interaction that gave a significantly different effect ( $P < 0.05$ ) between factor A (a type of solvent) and factor B (maceration time) to the percentage of total EKKBP catechins.

DMRT test results showed that P2M1 treatment (acetone solvent and maceration time 6 hours) had a significantly different effect on other treatments.

The percentage of total catechins obtained in this study ranged from 4.13% to 25.53%. The highest percentage of total catechins was obtained from the treatment of acetone with maceration of 6 hours, 25.53%, followed by an ethyl acetate solvent with a 6-hour maceration time of 23.68%. While the lowest catechin compounds were found in the treatment of water solvents at a temperature of 80 ° C with different maceration periods (6, 12, and 18 hours), respectively: 5.60%, 4.17%, and 4.13%.

In this study, it was found that the high yield did not reflect the high catechin value, this was because when extracted with polar or semi-polar solvents, other phytochemical compounds that matched the solvent also dissolved, causing a high total yield. According to (Achyadi *et al.*, 2018), the size of the yield cannot indicate the quality of the product, because the yield is of small value does not necessarily have a low-quality product, and vice versa, the yield of a large value is not necessarily the product has a low quality value. According to Vuong *et al.* (2010), the catechin extract obtained contained other components besides the catechins which were dissolved in the solvent as well as the residual solvent remaining, thereby affecting the high yield.

Types of solvents different in this study showed significant differences in the total catechin content of betel nut peel extract. The levels of catechins obtained from acetone solvents (19.87%) were not significantly different from those of ethyl acetate (19.75%) higher than those of catechins with water solvents at 80°C (4.90%). This is because acetone is an extraction solution that matches its polarity with the extracted compound, the areca nut peel catechin. This is supported by Sangthong *et al.* (2013) reported that acetone is the best solvent for extracting catechin compounds from areca seed. The choice of the type of solvent must consider the suitability of the solvent with the solute (polarity), toxicity, volatility, availability, and price of the solvent. Acetone is a polar solvent, liquid, colorless, food-grade (harmless if used as a solvent for food analysis) and flammable (Tiwari *et al.*, 2011).

In this study the longer the maceration time (6, 12 and 18 hours) the reduced levels of EKKBP catechins obtained for all types of solvents. The highest catechin levels were obtained at 6 hours maceration time (18.27%), then prolongation of maceration time to 12 hours, catechin levels were reduced at 12.91% and at 18 hours the catechin content level was 13.49%. The ability of a solvent to dissolve a compound is also influenced by the length of time of maceration. According to Maulida & Naufal (2014), the longer the extraction process, the longer the contact between the solvent and the solute, so that the solute dissolution process will continue until the solvent is saturated with the solute. However, in this study, an extended period of maceration showed lower levels of total catechins. According to Sintha (2008), the extraction time on each material has an optimum limit, the addition of time beyond its optimum limit has no effect because the compound that has moved to the solvent will experience a saturation point and has been extracted optimally.

#### IV. CONCLUSION

It concluded the combination of distillate acetone solvent and maceration duration for 6 hours was the best combination to obtain catechin extract from Areca catechu peel. In this condition the percentage of water content was 10.53%, yield percentage was 7.13%, and percentage of catechin extract was 25.53%.

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