



Optimized recombinant *Bacillus Subtilis* 168 whole-cell catalyzes one-step biosynthesis of high fructose syrup

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Received: 21 Nov 2020; Received in revised form: 11 Feb 2021; Accepted: 01 Mar 2021; Available online: 24 Mar 2021

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Abstract— High fructose syrup is a sweetener that can replace sucrose, which is widely used in the food and beverage industries. In this study, codon-optimized *Actinoplanes missouriensis* CICIM B0118 (A) derived glucose isomerase heterologous expression was realized in the food-grade strain *Bacillus Subtilis* 168, and the recombinant *Bacillus Subtilis* 168/pMA5-xylA was successfully constructed. The whole-cell bioconversion system of D-glucose isomerization to the biosynthesis of D-fructose was optimized. The results showed that the concentration of biocatalysts was DW 40 g/L, and the concentration of substrate D-glucose was 180 g/L, Mg²⁺ concentration 10 mmol/L, Co²⁺ concentration 1 mmol/L, continuous conversion at 70°C, 220 R/min for 18 h, D-fructose concentration reached 103.32 g/L, the conversion rate was 57.4%, realizing the high fructose syrup one-step safe biosynthesis. This research provided an experimental and theoretical basis for the industrialized and safe biosynthesis of high fructose syrup and had an important reference value.

Keywords— Glucose isomerase; isomerization; whole-cell catalysis; *Bacillus Subtilis* 168; high fructose syrup; biosynthesis.

I. INTRODUCTION

High Fructose Corn Syrup (HFCS), a purified, concentrated, an aqueous solution of functional saccharides currently used as a sugar substitute, is one of the most commonly used sweeteners in the production of nutritive beverages and foods, including soft drinks, ketchup, yogurts, ice-cream, chocolate milk, candies, jams, condiments, canned and packaged foods[1–3]. Besides, HFCS has been recently considered as a renewable resource for the production of 5-hydroxymethylfurfural and levulinic acid, which can be used in the synthesis of

other valuable biopetrochemicals: plastics, green solvents, lubricants, and valuable biofuels[4–8]. The health benefits of HFS intake have been proven by insulin-independent metabolism, increased absorption of iron and zinc, enhanced ethanol metabolism, low-sugar and low-calorie contents, and desirable organoleptic properties [2,9]. Other advantages of HFS compared with other types of sugars include high sweetness, high solubility, and low viscosity, flavor enhancement, good humectant, does not cause any side effects in acidic foods and doesn't form crystals[3,9,10].

According to the fructose content, HFCS can be divided into three types HFS-42, HFS-55, and HFS-90. However, due to the low fructose content of HFS-42, the medical and health care value cannot be brought into full play, and it is easy to crystallize and precipitate during low-temperature storage and transportation, so HFS-55 with higher fructose concentration has become the mainstream product[1,2]. The production methods of HFCS include chemical catalysis and enzymatic biocatalysis. The chemical catalytic method is that glucose is isomerized or acid hydrolyzed to fructose in an alkaline environment[2].

For more than a century, people have known that glucose can isomerize fructose through alkaline isomerization or acid hydrolysis. Therefore, this is a harsh process that leads to unacceptable sugar breakdown. Besides, the chemical synthesis of HFCS involves high calcination temperature and is not environmentally friendly[11–13]. The homogeneous Lewis acid generally has the problem of separation and recovery, and the synthesis of Sn- β zeolite is more complicated. On the other hand, Brønsted bases such as sodium hydroxide can also effectively isomerize glucose to fructose. In contrast, due to the severe degradation of fructose and glucose, they usually obtain a lower yield of glucose to fructose, and they also face the problem of separation and recovery[4,12,14].

Owing to the aforementioned drawbacks of environmentally unfriendly chemical synthesis, commercial HFCS has been mainly produced by microbial synthesis since the early 1970s after Yoshiyuki Takasaki discovered a thermo-stable glucose isomerase (GIase) enzyme from yeast [15]. GIase, also known as D-xylose isomerase, is a widely used enzyme for the production of HFCS. It catalyzes not only the conversion of D-glucose to D-fructose but also the conversion of D-xylose to D-xylulose [16]. With increasing HFCS consumption, the production level of GIase has drawn extensive attention in recent years. Many microbial strains, including *Streptomyces* sp. CH7[17], *Lactobacillus bif fermentans*[18], *Bacillus coagulans* [19], *Streptomyces murinus*, *Hyperthermophilic Thermotoga* [20], and *Pseudomonas hydrophila* [21], have been reported to produce glucose isomerase. Because of the increased global demand for HFCS, the level of GIase production has gained considerable attention, especially in the food and beverages industry. Due to the low productivity and stability of enzymes produced by wild-type microorganisms under harsh conditions, a more efficient expression system is needed for the production of recombinant GIase with desired properties for large-scale production of HFCS.

To obtain an effective expression mechanism, GIase has been heterologously expressed in a variety of hosts[22,23],

and a variety of fermentation techniques, including fed-batch and high-density fermentation, have been used. As a result, GIase expression has greatly improved; for example, Akdag et al.[24] announced that using a beet molasses-based feeding method, they achieved the highest recombinant GIase production, 35.3 U/mL, in *E. coli*. Due to its well-known genetics' history, short generation time, and suitability for low-cost high-density fermentation, *E. coli* is a common heterologous host for the expression of recombinant proteins[25]. However, improvements in GIase production, specially concerning HFCS-55 manufacturing, are still valued for industrial applications.

Although one-step biosynthesis of HFS-55 has achieved certain results, most of its host strains previously investigated with higher potential yields are *Escherichia coli* BL21, but in the catalytic process, BL21 may bring harmful toxin that do not meet the requirements of food safety into the target products[26–28]. Therefore, it is very meaningful to realize the heterologous expression of glucose isomerase in food safety strains with clear research background and to safely synthesize HFS-55 high fructose syrup in one step.

However, food safety requires a thorough investigation of food-grade microorganisms. *Bacillus subtilis* 168 and some other non-pathogenic related *Bacillus* species, which are free of exotoxins and endotoxins, and have a recognized history of safe use in foods are very useful for fermentation and large-scale cultivation[29].

Thus, an efficient expression of recombinant GIase in a generally regarded safe strain is necessary for improved economic HFCS-55 manufacturing and can contribute more to food security. In this study, a bacterial strain *Bacillus Subtilis* 168 was used as the host cell to heterologously express the glucose isomerase from *A. missouriensis* CICIM B0118 (A). The recombinant *Bacillus Subtilis* 168/pMA5-xylA was constructed. Using recombinant *Bacillus Subtilis* 168 as a whole-cell catalyst, the whole-cell catalytic conditions for isomerization of D-glucose to D-fructose were optimized. Under the optimal transformation system, HFS-55 high fructose syrup was synthesized safely in one step.

II. MATERIALS AND METHODS

2.1. STRAIN AND PLASMID

E. coli JM109, *Bacillus Subtilis* 168, and plasmid pMA5 were all preserved in our laboratory stock. The recombinant plasmid pET28a-xylA (genebank accession number: FJ858195.1) was synthesized by Suzhou Jinweizhi Biotechnology Co., Ltd.

2.2. EXPERIMENTAL REAGENTS

Restriction enzyme NdeI and MluI, high fidelity enzyme, and homologous recombination enzyme cloning kit were purchased from Takara company. Small Plasmid Extraction Kit, agarose gel DNA recovery kit, and so on are purchased from Shanghai Jarry Bioengineering Co., Ltd., chloramphenicol, isopropyl β -D- IPTG are purchased from Shanghai Bioengineering (Shanghai) Limited by Share Ltd, D- glucose is purchased from Mclean, D- fructose is purchased from Aladdin, glycerol, imidazole, sodium chloride and so on are all commercially analytical grades.

2.3. CULTURE MEDIUM AND CONDITIONS

LB medium: yeast extract 5 g / L, tryptone 10 g / L, sodium chloride 10 g / L (solid medium with 1.5% agar powder) for *E. coli* culture. The culture temperature was 37 ° C. TB medium: (12 g tryptone, 24 g yeast extract, 4 ml glycerol, 125.5 g K₂HPO₄, 23 g KH₂PO₄) per liter used for *B.subtilis*168 culture. The culture temperature was 30 ° C.

2.4. METHODS

2.4.1. CONSTRUCTION OF THE RECOMBINANT STRAIN

2.4.1.1. CONSTRUCTION OF RECOMBINANT *E. COLI* JM109/PMA5-XYLA

The recombinant plasmids pET28a-xylA and -xylA-F: ttattcagatgaaaacatatg atgagtgttcaagccaccg(NdeI), and -xylA-R: atttcgacctagaaacgct ttagegcgcaccacg(MluI) were used as plasmid template and primers respectively for PCR amplification. After ligation, the conjugated products were transformed into *E. coli* JM109 competent cells for cloning purpose, coated with LB plate containing 50 μ g / mL Kanamycin, and cultured overnight at 37 ° C incubator. The transformants were selected for colony PCR verification. The correct transformants were cultured in 10 ml liquid containing 50 μ g / mL Kanamycin for 8-12 h at 37 ° C and 180 R / min. Then, the strains were preserved and the recombinant plasmids were extracted. The extracted recombinant plasmids were sent to Suzhou Jinweizhi Biotechnology Co Ltd, for sequencing analysis. The sequencing results were analyzed by snap gene software, and the correct strain was named *E. coli* JM109 / pMA5- xylA, harboring the recombinant plasmid pMA5-xylA.

2.4.1.2. CONSTRUCTION OF RECOMBINANT *BACILLUS SUBTILIS* 168/PMA5-XYLA.

The recombinant plasmid pMA5-xylA was electroporated into *Bacillus Subtilis* 168 competent cells, coated on LB solid plate containing 50 μ g / mL Kanamycin, and cultured upside down in a 37 ° C incubator overnight. The correct

recombinant strains were named *Bacillus Subtilis* 168/pMA5-xylA

2.5. EXPRESSION OF GLUCOSE ISOMERASE (GI)

The preserved recombinant *Bacillus Subtilis* 168/pMA5-xylA was activated on LB plate containing 50 μ g / mL Kanamycin. The single colony was transferred to a 10 mL TB liquid medium (containing 50 μ g / mL kanamycin) and cultured at 30 ° C and 200 rpm for 12-24 h. Then, the 50 ml TB liquid medium (containing 50 μ g / mL kanamycin) was transferred to a 1% inoculation amount and cultured at 30 ° C and 200 rpm for 3-4 h. IPTG was added and cultured in a 30 ° C shaker for 12 h. The cells were centrifuged at 4 ° C and 10 000 rpm for 10 min. The cells were washed twice with PBS (pH 7.4, concentration 50 mmol / L) buffer and then suspended. After adding the appropriate amount of lysozyme, the cells were placed on ice for 2 ~ 3 h. The cells were broken by an ultrasonic crusher. The resultant solution was centrifuged for 20 min at 4 ° C and 12 000 R / min. The supernatant was used for SDS-PAGE analysis.

2.6. THE WHOLE-CELL CATALYTIC ACTIVITY OF RECOMBINANT *BACILLUS SUBTILIS* 168

Reaction system[30] (25 ml): 2.5 ml 8 mM MgCl₂, 2.5 ml 200 μ mol / L CoCl₂, 12.5 ml, 2 mol / L D-glucose and 1 g recombinant *Bacillus Subtilis* 168. The reaction was incubated at 70 ° C for 1 h and stopped ice bath for 5 min. The supernatant was centrifuged and analyzed by HPLC. The HPLC conditions were: RID differential detector, Hi-plex-Ca (300 mm \times 7.7 mm) column, ultrapure water as mobile phase, the flow rate of 0.6 ml/min, column temperature of 80 ° C, injection volume of 10 μ L, detection wavelength of 210 nm.

2.7. OPTIMIZATION OF WHOLE-CELL CATALYSIS CONDITIONS

2.7.1. EFFECT OF CELL CONCENTRATION ON WHOLE-CELL CATALYTIC SYSTEM

Under the conditions of substrate D-glucose concentration of 1 mol / L, pH 7.0, Mg²⁺ concentration of 10 mmol / L, Co²⁺ concentration of 1 mmol / L, DCW of 20, 30, 40, 50, and 60 g / L, respectively, the reaction time was 1 h at 70 ° C, and the D-fructose content in the transformation solution was detected by HPLC to determine the optimal bacterial concentration.

2.7.2. EFFECT OF METAL IONS ON WHOLE-CELL CATALYTIC SYSTEM

Under the conditions of cell concentration 40 g / L, substrate D-glucose concentration 1 mol / L, pH 7.0, different metal ions Ba²⁺ (1 mmol / L), Cu²⁺ (1 mmol / L), Fe²⁺ (1 mmol / L), Mg²⁺ (10 mmol / L), Ca²⁺ (1 mmol / L), Mn²⁺ (1 mmol / L), Co²⁺ (1 mmol / L) and Zn²⁺ (1 mmol /

L) were added respectively, and the reaction time was 1 h at 70 ° C. The content of D-fructose in the conversion solution was determined by HPLC.

2.7.3. EFFECT OF SUBSTRATE D-GLUCOSE CONCENTRATION ON WHOLE-CELL CATALYTIC SYSTEM

Under the conditions of DCW 40 g / L, pH 7.0, Mg²⁺ 10 mmol / L and Co²⁺ 1 mmol / L, the concentration of substrate D-glucose in the whole-cell transformation system was controlled to be 0.5 mol / L, 1.0 mol / L, 1.5 mol / L and 2.0 mol / L, and the reaction time was 1 h at 70 ° C. The content of D-fructose in the transformation solution was determined by HPLC.

2.7.4. EFFECT OF PH ON WHOLE-CELL CATALYTIC SYSTEM

Under the conditions of cell concentration DCW 40 g / L, substrate D-glucose 1 mol / L, Mg²⁺ 10 mmol / L and Co²⁺ 1 mmol / L, the reaction was carried out at pH 4.0-10.0 (interval 1.0) and 70 ° C for 1 h respectively. The content of D-fructose in the conversion solution was determined by HPLC to determine the optimal reaction pH. The buffer solutions included acetic acid sodium acetate buffer with pH 4.0, 5.0, and 6.0, Tris-HCl buffer with pH 7.0 and 8.0, glycine NaOH buffer with pH 9.0 and 10.0, and the concentrations were all 50 mmol / L.

2.7.5. EFFECT OF REACTION TEMPERATURE ON WHOLE-CELL CATALYTIC SYSTEM

Under the conditions of cell concentration DCW 40 g / L, substrate D-glucose 1 mol / L, pH 7.0, Mg²⁺ 10 mmol / L and Co²⁺ 1 mmol / L, the whole-cell catalytic system was placed in a constant temperature water bath at 40, 50, 60, 70 and 80 ° C for 1 h and the D-fructose content in the conversion solution was detected by HPLC to determine the optimal reaction temperature.

2.8. WHOLE-CELL BIOCATALYTIC BIOSYNTHESIS OF HIGH FRUCTOSE SYRUP

The preserved recombinant *Bacillus Subtilis* 168/pMA5-xylA was activated on an LB plate containing 50 µ g / ml kanamycin. The single colony was transferred to 10 ml TB liquid medium (containing 50 µ g / mL kanamycin) for 12 h at 37 ° C and 200 rpm/min, and then transferred to 200 ml TB liquid medium (containing 50 µ g / mL kanamycin) at 1% inoculum for 3-4 h at 37° C and 200 rpm/min. IPTG was added and cultured in a 30 ° C shaker for 12 h. After centrifugation for 10 min at 4 ° C and 10 000 rpm/min, the cells were collected and used as the whole-cell biocatalysts.

Under the optimal conditions of whole-cell transformation (cell concentration DCW 40 g / L, substrate D-glucose concentration 1 mol / L, pH 8.0, Mg²⁺ concentration 10

mmol / L, Co²⁺ concentration 1 mmol / L), the reaction was carried out on a magnetic stirrer at 70 ° C and 220 rpm for 18 h. 1 ml sample was collected every 3h, centrifuged at 4 ° C and 12000 R / min for 15 min, and the supernatant was analyzed by HPLC to detect the content of D-fructose.

III. RESULTS AND DISCUSSION

3.1. CLONING AND EXPRESSION OF XYLA GENE ENCODING GLUCOSE ISOMERASE

The gene sequence of glucose isomerase from *A. missouriensis* CICIM B0118 (A) was retrieved from the NCBI database [31], with a length of 1185 bp, which was submitted to Suzhou Jinweizhi Biotechnology Co Ltd for artificial synthesis, and the recombinant plasmid pET28a-xylA was obtained. The plasmid was used as a template for PCR amplification. The electrophoretic results are shown in Figure 1a. The specific bands (Lane 1 and 2 in Figure 1a) are consistent with the target. The recombinant *E. coli* JM109 / pMA5 -xylA was constructed by transferring the ligation product into *E. coli* JM109 competent cells. The recombinant plasmid pMA5-xylA obtained from the culture of recombinant *E. coli* JM109 /pMA5-xylA was transferred into *Bacillus subtilis* 168 to construct recombinant *Bacillus subtilis* 168/pMA5-xylA. The expression of GI was induced by IPTG and analyzed by SDS-PAGE. The results are shown in Fig. 1b. It can be seen from Lane 2 and 3 in the figure that the recombinant strain *Bacillus subtilis* 168/pMA5-xylA has an obvious protein expression band at the molecular weight of 43kDa as reported previously[30], that is, GI was successfully expressed in *Bacillus subtilis* 168.

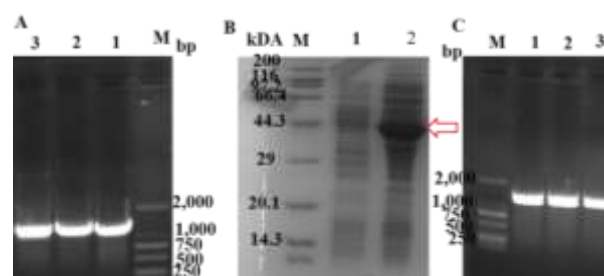


Fig. 1 Gene cloning and expression analysis of GI

Fig. A: PCR amplification of xylA gene, M: 2,000 BP nucleic acid marker; Fig. B: heterologous expression of glucose isomerase (GI), M: protein marker, lane 1: *B. subtilis* 168 wild type broken cells supernatant, lane 2: *B. subtilis* 168/pMA5-xylA broken cells supernatant; Fig. C: M:2,000 BP nucleic acid marker ,lane 1,2,& 3:*B. subtilis* 168/pMA5-xylA transformant colony PCR verification results.

3.2. WHOLE-CELL CATALYTIC PERFORMANCE TEST

Using whole-cell as a catalyst has gradually replaced crude enzyme or pure enzyme in the biosynthesis of target products because cells can protect the enzyme from the adverse environment and shear force, and batch conversion can be repeated, and no cofactor or coenzyme need to be added to the reaction process[32]. Therefore, the whole cell of *Bacillus subtilis* 168 will be used as a catalyst for D-glucose isomerization to synthesize D-fructose.

Firstly, the whole-cell catalytic performance of recombinant *Bacillus subtilis* 168 was tested to see whether it can isomerize D-glucose to D-fructose. Results as shown in Figure 2, the whole-cell biocatalyst of *Bacillus subtilis* 168 could isomerize D-glucose to produce D-fructose. In the figure, the peak time of D-glucose and D-fructose was 17.106 min and 21.380 min respectively, which indicated that *Bacillus subtilis* 168 had the potential to produce high fructose syrup. The next step is to optimize the whole-cell catalytic system of D-glucose isomerization to D-fructose, and explore the effects of cell concentration, divalent metal ions, substrate D-glucose concentration, reaction pH, and reaction temperature on D-fructose synthesis.

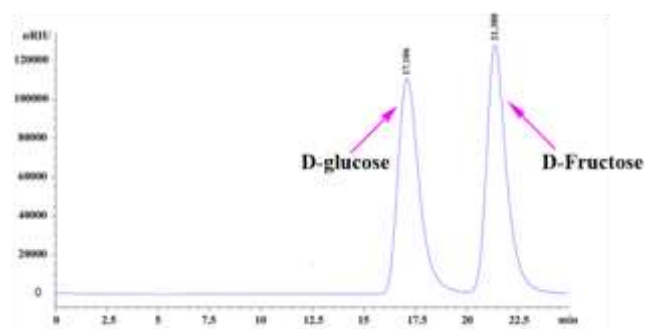


Fig. 2 HPLC detection results of the conversion solution

3.3. OPTIMIZATION OF WHOLE-CELL CATALYSIS CONDITIONS

3.3.1. EFFECT OF CELL CONCENTRATION ON BIOTRANSFORMATION OF D-GLUCOSE TO D-FRUCTOSE

The cell concentration reflects the content of glucose isomerase GI in the conversion system to a certain extent. Therefore, the different cell concentrations may affect the catalytic process of D-glucose isomerization to D-fructose, as shown in Figure 2. With the increase of cell concentration, D-fructose content increased; when cell concentration exceeded 40 g / L, although substrate concentration increased, D-fructose content decreased. When the cell concentration DCW was 40 g / L, the D-

fructose content was the highest, which was 3.1 g / L. the optimal cell concentration for whole-cell catalytic synthesis of D-fructose was 40 g / L.

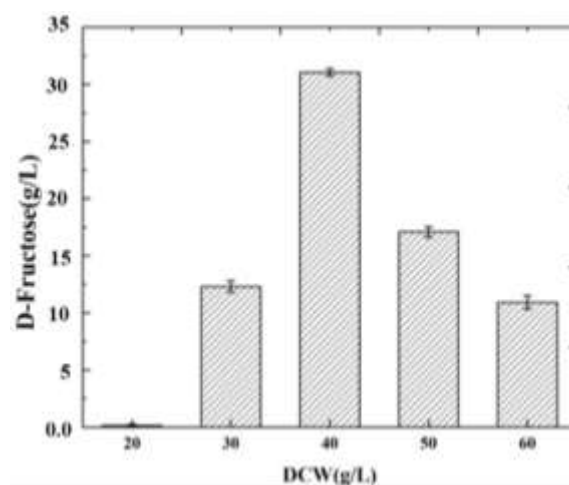


Fig. 3 The effect of cell concentration on D-glucose biotransformation to D-fructose

3.3.2. EFFECTS OF DIVALENT METAL IONS ON BIOTRANSFORMATION OF D-GLUCOSE TO D-FRUCTOSE

GI is a metal enzyme, combined with divalent metal ions, in the process of D-glucose biotransformation to D-fructose, metal ions are needed to assist the isomerization reaction, and divalent metal ions are very important for the activity and stability of GI [25]. Different GI needs different divalent metal ions. Most of the reported GI mainly use Mg^{2+} , Mn^{2+} , Co^{2+} [33], or two metal ions as auxiliary catalysts [24,33].

Different divalent metal ions may have different effects on the catalytic process of D-glucose isomerization to D-fructose. Therefore, the effects of different divalent metal ions Ba^{2+} (1 mm), Cu^{2+} (1 mm), Fe^{2+} (1 mm), Mg^{2+} (10 mm), Ca^{2+} (1 mm), Mn^{2+} (1 mm), Co^{2+} (1 mm) and Zn^{2+} (1 mm) on D-glucose isomerization to D-fructose were tested in this study. The results are shown in Fig. 3. The results showed that the presence of divalent metal ions Cu^{2+} , Ca^{2+} , and Zn^{2+} was not conducive to the isomerization of D-glucose to produce D-fructose. The divalent Cu^{2+} , Ca^{2+} , and Zn^{2+} exerted the inhibition effect on the activity of GI, so it was not conducive to the synthesis of D-fructose. The presence of bivalent metal ions Ba^{2+} , Fe^{2+} , Mg^{2+} , Mn^{2+} , and Co^{2+} promoted the synthesis of D-fructose, especially the presence of Mg^{2+} and Co^{2+} , which made the D-fructose content reach 4.95 g / L and 2.96 g / L respectively, 4.85 and 2.9 times of the control (1.02 g / L).

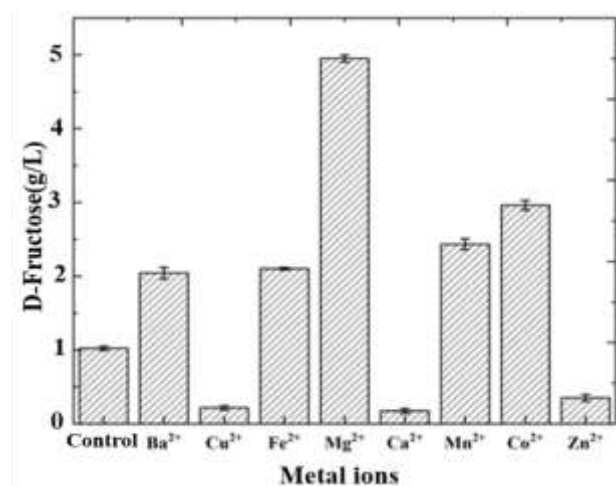


Fig. 4 The effect of divalent cations on the biotransformation of D-glucose into D-fructose

3.3.4. EFFECT OF SUBSTRATE D-GLUCOSE CONCENTRATION ON THE BIOTRANSFORMATION OF D-GLUCOSE TO D-FRUCTOSE

An appropriate increase of substrate concentration can accelerate the reaction rate, which is conducive to product synthesis. It can be seen from Fig.5 that the D-fructose content increases with the increase of D-glucose concentration when the substrate D-glucose concentration is in the range of 0 ~ 1.0 mol / L. when the substrate D-glucose concentration is 1.0 mol / L, the D-fructose content reaches the highest, 27 g / L. When the concentration of substrate was higher than 1.0 mol / L, the concentration of D-glucose increased, but the content of D-fructose decreased. It may be that if the concentration of D-glucose exceeded a certain value, the activity of the GI enzyme would be inhibited, which was not conducive to the isomerization of glucose to D-fructose. Therefore, the content of D-fructose decreased.

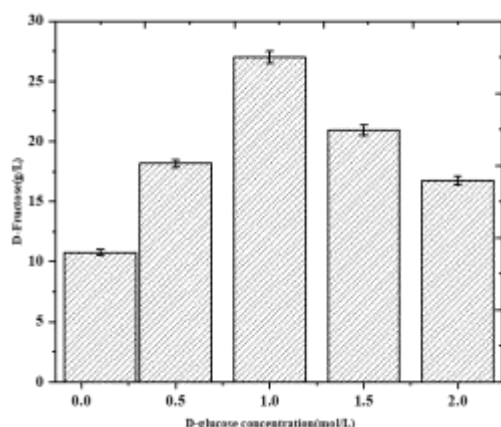


Fig. 5 The effect of substrate D-glucose concentration on D-glucose biotransformation to D-fructose

3.3.5. EFFECT OF PH ON BIOTRANSFORMATION OF D-GLUCOSE TO D-FRUCTOSE

In the process of isomerization of D-glucose to D-fructose, the optimal pH is generally 7.0 ~ 9.0, and pH has a significant effect on the biotransformation rate of D-fructose. A few GI have good catalytic activity in weak acid pH environments, while isomerization under weak acid conditions can reduce the formation of by-products [35], [35]. It can be seen from Fig. 6 that the GI from codon-optimized *A. missouriensis* CICIM B0118 (a) in this study is at a high level in the range of pH 7.0 ~ 9.0, reaching 44.83 g / L, 48.10 g / L, and 46.56 g / L, respectively. However, under acidic pH 4.0 ~ 6.0 and peralkaline pH 10.0, the content of D-fructose was not high, which was not conducive to the isomerization of D-glucose to D-fructose.

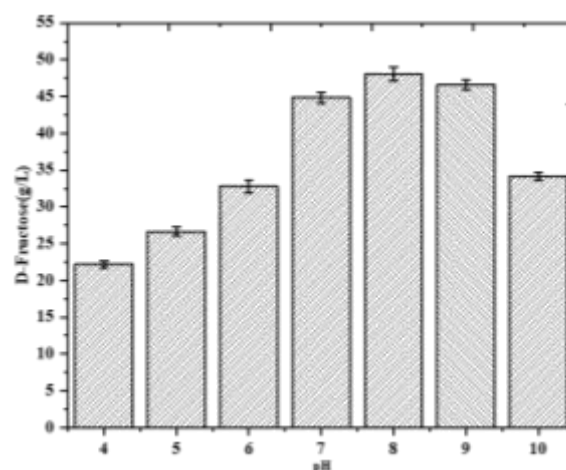


Fig. 6 The effect of pH on D-glucose biotransformation to D-fructose

3.3.5. EFFECT OF REACTION TEMPERATURE ON BIO-TRANSFORMATION OF D-GLUCOSE TO D-FRUCTOSE

Previous studies have shown that the isomerization of D-glucose catalyzed by GI is a thermodynamic equilibrium reaction. With the increase of temperature, the catalytic equilibrium moves towards the formation of D-fructose, and high temperature is conducive to the formation of D-fructose [1,10]. Therefore, this study also studied the reaction temperature of the whole-cell catalytic system for D-glucose isomerization to D-fructose, and the results are shown in Fig.7. It can be seen from the figure that GI from *A. missouriensis* CICIM B0118 (a) can tolerate a wide range of temperatures 70 °C is the optimal temperature for biotransformation. At this time, the D-fructose content is 50.34 g / L; at 90 °C, the D-fructose content is still at a high level, 36.23 g / L, only 16.22% lower than the

optimal temperature. Due to the rapid decrease of enzyme activity at 80 °C in Sweetzyme®, a commercial enzyme, the content of D-fructose decreased critically. In contrast, GI from *A. missouriensis* CICIM B0118 (A) has higher catalytic activity at high temperatures, which is beneficial to the industrial production of high fructose syrup.

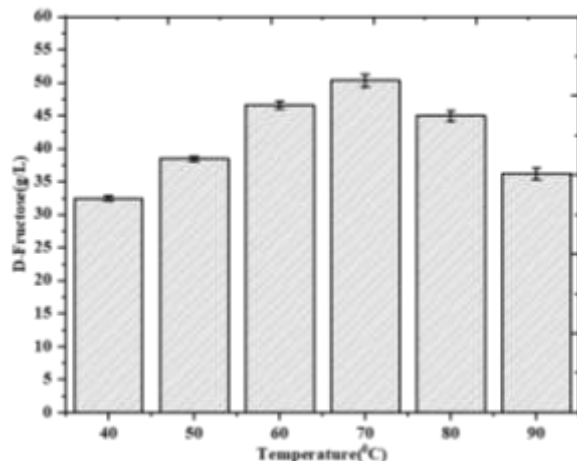


Fig. 7 The effect of reaction temperature on the biotransformation of D-glucose to D-fructose

3.3.6. WHOLE-CELL CATALYTIC SYNTHESIS OF HIGH FRUCTOSE SYRUP

Under the optimal transformation conditions, 4 g DWC were added to 100 ml Tris-HCl buffer (pH 7.0) of 180 g / L glucose, 10 mmol / L Mg^{2+} and 1 mmol / L Co^{2+} , and transformed at 70 ° C and 220 rpm. The supernatant was centrifuged every 3 h, and the content of D-fructose was determined by HPLC. Results as shown in Fig. 8, in the early stage of transformation (0 ~ 15 h), D-fructose concentration gradually increased with the consumption of substrate D-glucose. At 15h, the remaining D-glucose concentration in the transformation solution was 71.87 g / L, D-fructose concentration reached 103.23 g / L, and the conversion rate was 57.35%. After that, D-glucose consumption slowed down, the reaction basically tended to balance, and D-fructose concentration almost remained constant, 18.5% 23 g / L, D-fructose concentration reached 103. 32 g / L and the conversion rate was 57. 4%. It can be seen that the recombinant strain *B.subtilis* 168/ pMA5-xylA can safely synthesize HFCS-55 high fructose syrup in one step, which provides an important reference for the sustainable and safe industrial production of HFCS-55 high fructose syrup.

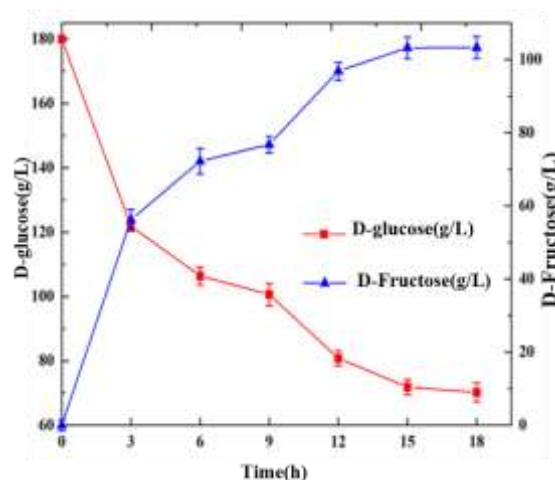


Fig. 8 Whole cells catalyze the isomerization of D-glucose to D-fructose

IV. CONCLUSION

In this paper, the heterologous expression of GI from *A. missouriensis* CICIM B0118 (A) in food safety bacterial strain *Bacillus subtilis* 168 was successfully realized for the first time, and the recombinant *Bacillus subtilis* 168 /pMA5-xylA was constructed to construct the recombinant *Bacillus subtilis* 168 pMA5-xylA and was used for the whole-cell biocatalysis, the whole-cell transformation conditions (cell concentration, a divalent cation, substrate concentration, pH and temperature) of D-glucose isomerization to D-fructose were optimized. Under the optimal transformation conditions, the concentration of D-fructose reached 103.32 g / L and the conversion rate was 57.4% for 18 h. One-step safe biosynthesis of HFCS-55 was realized. It is found that GI from *A. missouriensis* CICIM B0118 (A) has good activity in a wide range of temperatures and pH, respectively 60 ~ 85 ° C and pH 7.0 ~ 9.0, which are suitable for industrial production of HFCS-55 high fructose syrup. The isomerization of D-glucose to D-fructose needs the assistance of divalent metal ions. Divalent metal ions have three functions on GI: activating, stabilizing, and improving the affinity of the D-glucose- GI enzyme. The next step will focus on the effect of divalent metal ions on the biotransformation of D-fructose to realize the sustainable and safe industrial production of high fructose syrup, we should improve the enzyme activity of GI and accelerate the biotransformation rate.

ACKNOWLEDGMENT

This work was funded by the National Key Research and Development Program of China (2018YFA0900300), the National Natural Science Foundation of China (Grant Nos. 21778024, 31870066, 31570085), National First-Class

Discipline Program of Light Industry Technology and Engineering (Grant No. LITE 2018-06), Fundamental Research Funds for the Central Universities (Grant No. JUSRP51708A), the 111 Project (Grant No. 111-2-06), and the Priority Academic Program Development of Jiangsu Higher Education Institutions.

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